

# Adsorption for wastewater treatment and water purification

## Part 2: Low-cost adsorbents for wastewater treatment applications

### Circular Economy for Water



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21.9.2020

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# Content and questions considered:



Illustration: Tatiana Samarina

- ✓ Conventional and emerging pollutants
- ✓ Adsorbents characteristics and adsorption capacities
- ✓ Methods and techniques applied
- ✓ Source of low-cost raw materials for adsorbent production
- ✓ Simplest variations of layouts with adsorption



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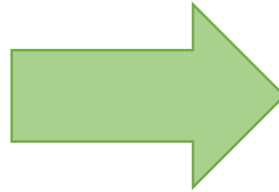
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# Conventional and emerging pollutants

Deteriorating of natural waters:

- anthropogenic activities
- population growth
- unplanned urbanization
- rapid industrialization
- unskilled utilization of water sources



## Conventional pollutants

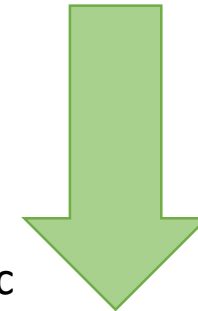
Nutrients – forms of phosphorous and nitrogen;  
Dyes and surfactants;  
Heavy metals;  
Bacteria and viruses  
Compounds polycyclic aromatic hydrocarbons.

Read more:

[https://www.sciencedirect.com/science/article/pii/S2095633915000039#:~:text=Emerging%20pollutants%20\(EPs\)%20are%20defined,\(or\)%20human%20health%20effects.](https://www.sciencedirect.com/science/article/pii/S2095633915000039#:~:text=Emerging%20pollutants%20(EPs)%20are%20defined,(or)%20human%20health%20effects.)

<https://www.sciencedirect.com/science/article/pii/B978044453199500052X>

## Emerging pollutants



Biorecalcitrant compounds, micro/nanoplastic compounds of drugs, pharmaceuticals and personal care products, plasticizers, fire retardants, endocrine disrupting chemicals, etc.



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# Low-cost adsorbents in water remediation techniques:

## Cost-effective adsorbents for nutrient removal

### Abundant and low cost source

Natural zeolites and clays  
Industrial wastes and by-products  
Agricultural residues  
Bio-based materials

### Manufacturing and remediation process

Ease of handling  
Low energy consumption  
Low GHG emission  
High capacity and short removal time

### Zero-waste Waste produced

Regeneration  
Recycling  
Reuse in other application

## Commercial adsorbents:

Activated carbons (PAC/GAC)  
1200-3000 EUR/ton  
GHG emission; energy demand;  
loss of adsorbent on reg.stage;  
utilization problems

Ion-exchange resins (polymers)  
1700-3000 EUR/ton  
energy demand; organic solvents;  
utilization problems

Zeolites (natural minerals)  
600-1500 EUR/ton  
unsteady quality; limit abundances

Sands, gravels, etc.  
50-350 EUR/ton  
low capacity and nonspecific treatment



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Read more:

<https://www.tandfonline.com/doi/abs/10.1080/10643380801977610>



# Sources of low-cost raw materials for adsorbent production

- ❖ agricultural and household wastes- e.g. fertilizer industry waste;
- ❖ industrial by-products - fly ash, steel industry wastes, aluminium industry wastes;
- ❖ sludges;
- ❖ sea materials - chitosan and seafood processing wastes, seaweed and algae;
- ❖ soil and ore materials - clays, zeolites, sediment and soil, and ore materials;
- ❖ novel low-cost adsorbents - peat moss, other industry waste such as those from leather industry and paper industry.



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# Adsorbents characteristics and adsorption capacities

**Adsorption** - process that allow to collect one of the component (**adsorbate**) contained in liquid, for instance in mine effluent, on the solid surface (**adsorbent**).

To describe particular behavior of adsorbate on chosen adsorbent, an equilibrium model is used. Such model is called isotherm of adsorption.

There are plenty of those have been proposed, but the main their application is to express the amount of adsorbate to be adsorbed per unit of adsorbent as a function of concentration.



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# Isotherms could be different forms, which reflects processes and limitation steps

## Isotherms:

- Need to be determined for particular stream treated
- Are used for calculations and design of treatment systems

## Read more:

N. Ayawei, A. N. Ebelegi, and D. Wankasi, "Modelling and Interpretation of Adsorption Isotherms," *J. Chem.*, vol. 2017, p. 3039817, Sep. 2017, doi: 10.1155/2017/3039817.  
<https://www.hindawi.com/journals/jchem/2017/3039817/>

Mahmoud, Dalia Khalid et al. "Langmuir model application on solid-liquid adsorption using agricultural wastes: Environmental application review." (2012).  
[https://www.academia.edu/3042397/Langmuir\\_model\\_application\\_on\\_solid\\_liquid\\_adsorption\\_using\\_agricultural\\_wastes\\_Environmental\\_application\\_review](https://www.academia.edu/3042397/Langmuir_model_application_on_solid_liquid_adsorption_using_agricultural_wastes_Environmental_application_review)



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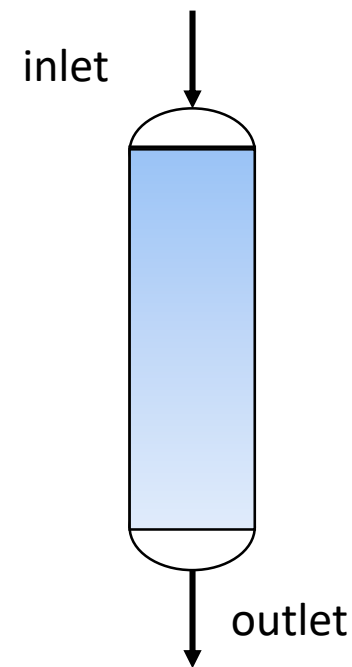
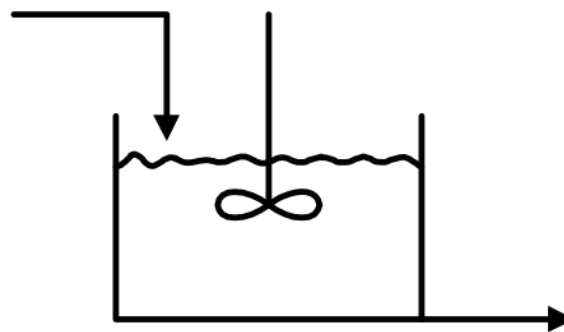
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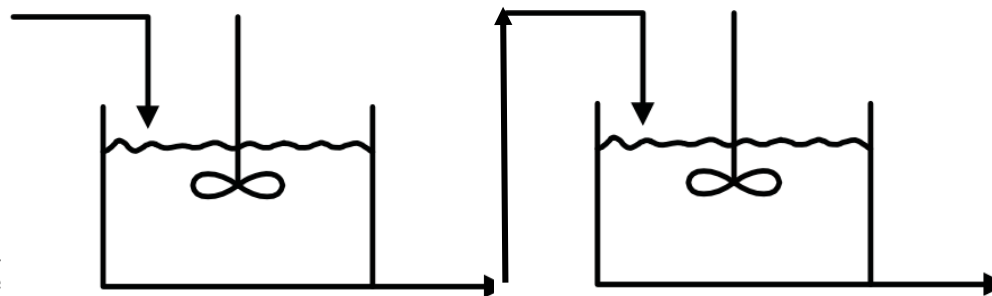
# Simplest variations of layouts for wastewater treatment

a completely mixed flow reactor (CMFR)

a column contactor



Both variants could be used in the cascade



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# Example of process calculation

An mine wastewater contains 10 mg/L of undesired contaminant, and is going to be treated by adsorption. 95% removal is required to reach safe discharge limit. The wastewater is discharged at a rate of 100 000 L/day. Calculate the mass of adsorbent requirement for treatment in a **completely mixed flow reactor**, if Freundlich isotherm given

$q = 5.1C^{0.87}$ , the  $q$  is adsorption capacity, mg/g and  $C$  is concentration, mg/L.

- Calculating the capacity of adsorbent at desired concentration (0.5 mg/L, 95% removed):

$$q = 5.1 * 0.5^{0.87} = 5.1 * 0.547 = \mathbf{2.79 \text{ mg adsorbat/ g adsorbent}}$$

- Calculating of contaminant load

$$Q = (C_0 - C_m) * \text{rate (L/day)} = 9.5 \text{ mg/L} * 100000 \text{ L/day} = 950 \text{ g/day}$$

- Mass of adsorbent needed

$$M = Q/q = 950 \text{ 000 mg/day} : 2.79 \text{ mg/g} = \mathbf{340.5 \text{ kg/day}}$$



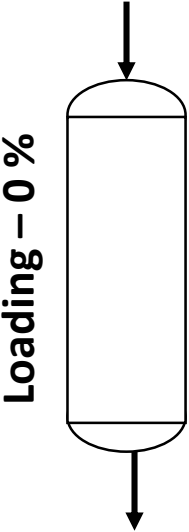
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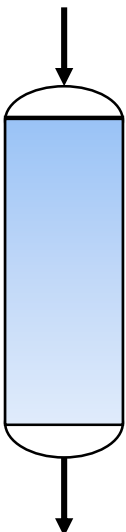
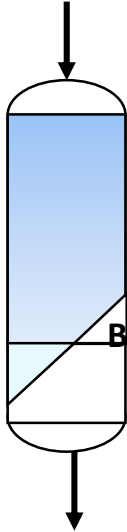
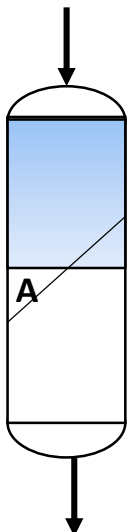
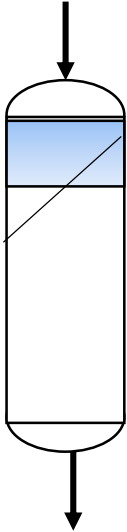


# Active zones at various times during adsorption and the breakthrough curve

This zone moves downward with a constant velocity as the upper regions become saturated.



Loading – 0 %

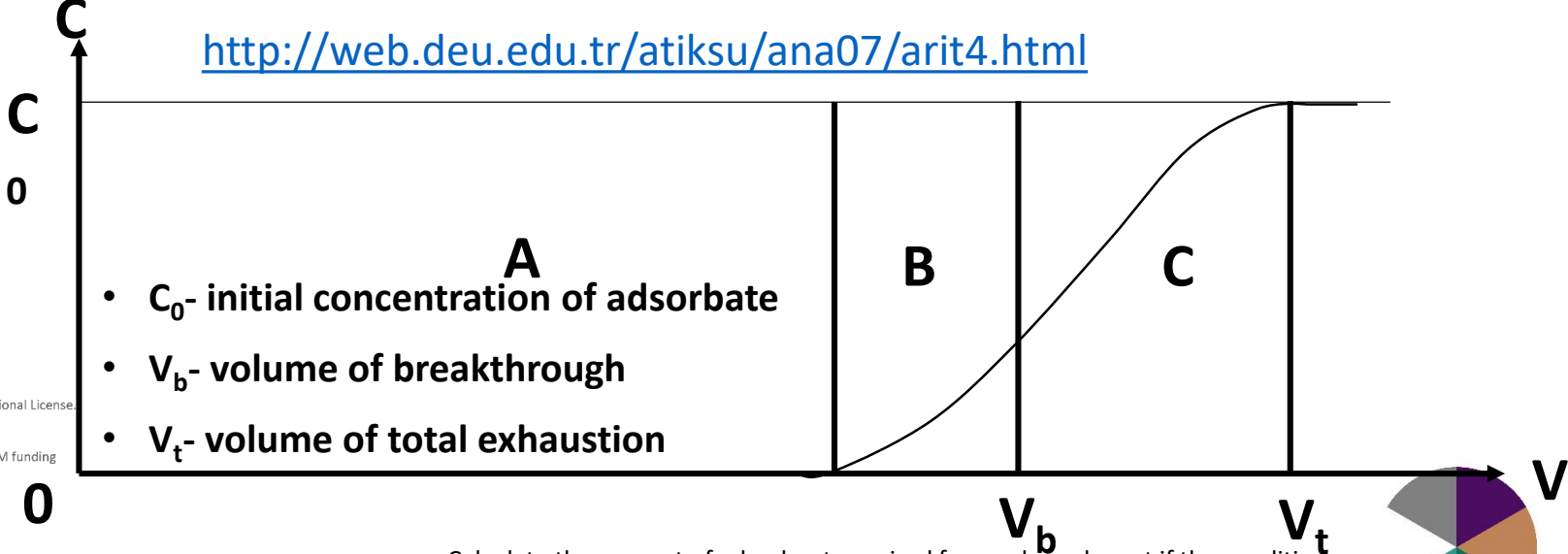


Loading – 100 %

Mass transfer zone, MTZ, part of a column where the solute is most effectively and rapidly adsorbed .

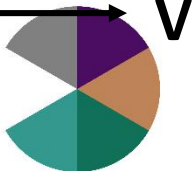
To design a column accurately, a test column **breakthrough curve** for the stream of interest and the chosen adsorbent need to be used.

<http://web.deu.edu.tr/atiksu/ana07/arit4.html>



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# Content bullets and conclusions:

- Adsorption is powerful tool for water purification and waste management;
- There is a need to develop more efficient selective, inexpensive and eco-friendly low cost adsorbents;
- Low cost adsorbents can be used for wastewater management in small communities and remote areas;
- Continuous process can be used for adsorption process;
- Regeneration of adsorbents could decrease the overall expenditures of technology



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